Periodic Table

| Name | _ Period |
|--|--|
| Directions: Use a Periodic table to find the | e information asked for below: |
| 1. What is the atomic number of: Calcium Iron Gold Uranium 1. What is the atomic number of: Calcium Iron Uranium In the atomic number of: Calcium Uranium In the atomic number of: In the atomic number of: Calcium Uranium In the atomic number of: In the atomic number of: Calcium Uranium In the atomic number of: Calcium | 2. What is the Atomic mass of: Calcium Iron Uranium Copper |
| 3. How many protons do the following have? Calcium Gold Copper Iron | |
| 4. How many electrons do the following have? Gold Iron Copper Uranium General Control Co | · , |
| 5. Does mercury have more protons and electrons the | han tin? |
| 6. Is mercury a heavier element than tin? | * |
| 7. Does potassium have more electrons than neon? | |
| 8. Does hydrogen have more electrons than Uraniur | m? . |
| 9. Which has more protons, sulfur or iodine? | • |
| 10. Write the symbols or the names for each of these Chlorine | e elements: Zn |
| Copper | Helium |
| Potassium | Iron |
| Silver | P |
| Na | Ne |
| Sn | Mercury |